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# METN 201-001: Intro Principles of Materials Engineering

Kathleen McEnnis

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# MTEN 201 - 001 Introductory Principles of Materials Engineering Fall 2024 Syllabus

**Instructor:** Dr. Kathleen McEnnis

PhD, Assistant Professor in CME Department

She/Her/Hers pronouns Email: mcennis@njit.edu Office: 382 Tiernan Hall

Zoom Room:

https://njit-edu.zoom.us/j/2945567196?pwd=OGlQcVZLeUlycFM2MEZvcWtaQk1OZz09 Office Hours: Wednesdays 1:30-2:30pm (in person or virtual) and Thursdays 1:30-2:30pm

(virtual). Schedule an appointment through email. Please email me for

other times.

**Teaching Assistant**: Shima Baloochestanzadeh (sb2936@njit.edu)

**Required eTextbook (Online Subscription Required):** WileyPLUS: Fundamentals of Materials Science and Engineering: An Integrated Approach, 6<sup>th</sup> Edition, W.D. Callister, Jr., and D.G. Rethwisch, John Wiley and Sons, Inc.

ISBN:

9781119764816 ~ WileyPLUS one-term access

or

9781119750499 ~ WileyPLUS one-term access + loose leaf text

Class: Monday 10:00 AM-11:20 AM Room FMH 409 (Face-to-Face) Friday 10:00 AM-11:20 AM Room FMH 409 (Face-to-Face)

**Course:** This course introduces the basic concepts of Materials Engineering, and covers introductory topics including structure, property, performance, and processing of materials. This course focuses on conventional materials including metallic materials and their alloys, ceramics, polymers, and composites. Relationship between structure and material properties, such as mechanical, electronic, thermal, optical, magnetic, and electrochemical, are investigated with a particular interest on ways to engineer material structures to produce desired set of properties. Broader themes associated with the property, processing and performance of materials that influence the economy, environment, and society are discussed.

Prerequisites: CHEM 126, PHYS 121 or PHYS 122, MATH 112.

Withdraw Deadline: November 11, 2024

**Course Administration:** Administration of this course will be done through Canvas where links to the WileyPLUS platform can be found.

**Assignments:** Homework assignments will be given using WileyPLUS or otherwise posted on Canvas. In class practice problems will also occasionally be assigned and will contribute to the assignment grade. Homework assignments (and practice problems) are graded and will be 10% of the overall grade. While the in-class practice problems can be worked on with other students, the WileyPlus assignments must be completed independently.

**Quizzes:** Quizzes will be given in class. Quizzes are closed book & notes, though an equation sheet will be provided. Quizzes are 20% of the total grade and the lowest quiz grade will be dropped.

**Exams:** There will be two exams and one final. The exams are each worth 20% of the total grade and the final is worth 30% of the total grade. All exams and finals will be closed book & notes, though an equation sheet will be provided.

#### **GRADING**

Assignments 10%
Quizzes 20%
Exams 40%

Exam 1 20% Exam 2 20%

Final <u>30%</u> 100%

Grades will be based on:

A: 90 – 100% B+: 85 – 89% B: 80 – 84% C+: 70 – 79% C: 60 – 69% D: 50 – 59% F: 0 – 49%

**Makeup Policy:** No makeup exams, finals, or quizzes will be granted unless the Dean of Students contacts me about your reason for missing and the reason is deemed suitable.

**Late Work Policy:** Assignments will not be accepted late unless there is an extenuating circumstance documented through the Dean of Student's office.

**Electronic Device Policy**: With the exception of calculators, the use of electronic devices during exams, finals, or quizzes is prohibited. The use of an electronic device during class time is allowed and you are encouraged to bring a personal electronic device to class to participate in PollEverywhere questions and access your WileyPLUS materials. Please be aware, however, that though these devices can aid in your learning experience, they can also be a source of distraction for both you and your peers. Use your electronic devices responsibly so as not to distract yourself or others from the class.

**Academic Integrity Policy:** Academic Integrity is the cornerstone of higher education and is central to the ideals of this course and the university. Cheating is strictly prohibited and devalues the degree that you are working on. As a member of the NJIT community, it is your responsibility to protect your educational investment by knowing and following the academic code of integrity policy that is found at:

http://www5.njit.edu/policies/sites/policies/files/academic-integrity-code.pdf.

Please note that it is my professional obligation and responsibility to report any academic misconduct to the Dean of Students Office. Any student found in violation of the code by cheating, plagiarizing or using

any online software inappropriately will result in disciplinary action. This may include a failing grade of **F, and/or suspension or dismissal from the university.** If you have any questions about the code of Academic Integrity, please contact the Dean of Students Office at <a href="mailto:dos@njit.edu">dos@njit.edu</a>.

Use of "homework help" sites such as Chegg.com to complete class work is prohibited. Any student found to have used one of these sites on an assignment will be reported to the Dean of Students Office for a potential academic integrity violation.

#### **Course Objectives**: Students will be able to:

- 1. name types of bonding in materials and identify which materials exhibit each of these bonding types
- 2. identify the structures of metals and ceramics
- 3. identify the chemical and structural characteristics of polymers
- 4. define types of imperfections and the roles they play in affecting the behavior of materials
- 5. design structures/components using predetermined mechanical properties and mechanical constraints and/or safety constraints
- 6. identify deformation and strengthening mechanism for materials
- 7. interpret phase diagrams for determination of phases present and computation of phase compositions and amounts
- 8. apply phase diagrams for development of microstructures in alloys
- 9. apply phase transformations concepts to determine microstructure
- 10. draw relationship between microstructure and mechanical properties
- 11. select appropriate materials for various applications
- 12. identify economic, environmental and societal issues in materials design and selection

#### **Topics & Dates** (may be subject to change):

1.	Introduction	Chapter 1
2.	Atomic Structure & Interatomic bonding	Chapter 2
3.	Metallic/Ceramic Structures	From Chapter 3

#### Quiz 1 Monday, September 23

4	Polymer Structures & Polymer Cryst.	Melting Tg	From Chapter 4 & 11
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#### Exam 1 Monday, October 7

5.	Imperfections in Solids	From Chapter 5
6.	Mechanical Properties	Chapter 7

### Quiz 2 Monday, October 21

7.	Deformation/Strengthening Mechanism	Chapter 8
8.	Phase Diagrams	Chapter 10

#### Exam 2 Monday, November 4

9. Phase Transformations Chapter 11

#### Quiz 3 Monday, November 18

10. Types and Applications of Materials

From Chapter 13

#### Quiz 4 Monday, Dec 2

11. Introduction to Fabrication of Materials12. Economic, Environmental and Social Issues

From Chapter 14

Chapter 20

Final Exam TBA (will take place during the Dec 15 - 21 Final Exam period)

#### **Disability Support Services**

If you need accommodations due to a disability please contact Chantonette Lyles, Associate Director of Disability Support Services, Fenster Hall Room 260 to discuss your specific needs. A Letter of Accommodation Eligibility from the Disability Support Services office authorizing your accommodations will be required.