

Fall 2020

CHEM 243-001: Organic Chemistry I

Yuanwei Zhang

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Chemistry: *Fall 2020 Course Syllabus*

NJIT Academic Integrity Code: All Students should be aware that the Department of Chemistry & Environmental Science (CES) takes the University Code on Academic Integrity at NJIT very seriously and enforces it strictly. This means that there must not be any forms of plagiarism, i.e., copying of homework, class projects, or lab assignments, or any form of cheating in quizzes and exams. Under the University Code on Academic Integrity, students are obligated to report any such activities to the Instructor.

COURSE INFORMATION

Course Description: This course offers students the opportunity to learn the nature of carbon in organic compounds. It presents general principles of organic chemistry related to nomenclature, structure, stereochemistry, uses and synthesis.

Number of Credits: 3

Prerequisites: General Chemistry

Course-Section and Instructors

Course-Section	Instructor
CHEM 243 - 001	Yuanwei Zhang

Office Hours: Mondays, 2:00 - 3:00 pm

Required Textbook:

Title	Organic Chemistry
Author	L. G. Wade Jr.
Edition	9 th edition
Publisher	Prentice Hall
ISBN #	032197137X

University-wide Withdrawal Date: The last day to withdraw with a **W** is Monday, November 09, 2020. It will be strictly enforced.

Learning Outcomes:

1. Identify sigma and pi bonds and explain the hybridization of the molecules
2. Discuss electronegativity and bond polarity
3. Draw Lewis structures, condensed structures and structural formulas of organic compounds
4. Use VSEPR to predict the geometry and polarity of molecules
5. Discuss resonance and delocalization of charge in molecules
6. Identify various functional groups in organic molecules, particularly alkenes, alkynes, alcohols, acids, ethers, esters, aldehydes, ketones and amines
7. Explain Lewis acid-base theory
8. Explain rules of nomenclature to describe the various hydrocarbons
9. Describe structural and geometric isomerism and the role of isomerism in determining molecule structure
10. Describe types of intermolecular forces
11. Apply knowledge of intermolecular forces to describe trends in boiling points and melting points of various molecules
12. Distinguish between conformers and isomers
13. Describe key reactions of alkanes (substitution), alkenes and alkynes (addition)
14. Identify chair and boat conformations of cyclohexane and predict the stability of the different conformers
15. Recognize the steps involved in free radical reactions
16. Identify and describe the reactivity of various intermediates produced during reactions of hydrocarbons
17. Propose reaction mechanisms to describe product formation in simple alkane substitution reactions
18. Describe chirality and distinguish between R and S stereoisomers and diastereoisomers
19. Discuss and draw the products of substitution and elimination reactions of alkyl electrophiles
20. Describe substitution reactions and recognize SN1 and SN2 reactions
21. Explain E1 and E2 reactions
22. Describe various types of addition reactions in aliphatic alkenes and alkynes
23. Describe properties, synthesis and reactivity of alcohols
24. Explain various organic reagents like Grignard reagents and borohydrides and their uses in specific reactions

POLICIES

All CES students must familiarize themselves with, and adhere to, all official university-wide student policies. CES takes these policies very seriously and enforces them strictly.

Grading Policy: The final exam will be cumulative. Before each Exam there will be a Quiz, which covers three or four chapters. Take the Quizzes seriously as they add up to one Exam grade. Problems in the body of the chapter are assigned and selected problems at the end of the chapter. These will not be collected. To do well in the course it is important to do these problems. Failure to take the exam in the assigned room will result in a loss of a full grade from your exam score. The final grade in this course will be determined as follows: The grade will be determined from a total of 300 points. 100 points will be dropped from the total of 400 points, either the quizzes (100 points) or the lower of the two midterm exams (The final exam cannot be dropped). Makeup exams are not encouraged. If you must miss an exam contact the coordinator before the exam or immediately after. Makeup should be taken within the first week of the exam and before exams are given back. You must have a valid excuse and a doctor note. The final grade in this course will be determined as follows:

The final grade in this course will be determined as follows:

Quizzes	100
Midterm Exam I	100
Midterm Exam II	100
Drop the lowest	-100
Final Exam	100

Your final letter grade in percentage in this course will be based on the following tentative curve:

A	90 - 100	C	70 - 74.9
B+	85 - 89.9	D	60 - 70
B	80 - 84.9	F	< 60
C+	75 - 79.9		

Attendance Policy: Each class is a learning experience that cannot be replicated through simply “getting the notes.”

Homework Policy: Homework is an expectation of the course.

Exams: There will be two midterm exams held in class during the semester and one comprehensive final exam. The following exam periods are tentative and therefore possibly subject to change:

Midterm Exam I	October 09
Midterm Exam II	November 10
Final Exam Period	December 15 - December 21

The final exam will test your knowledge of all the course material taught in the entire course.

Makeup Exam Policy: There will normally be **NO MAKE-UP QUIZZES OR EXAMS** during the semester. In the event that a student has a legitimate reason for missing a quiz or exam, the student should contact the Dean of Students office and present written verifiable proof of the reason for missing the exam, e.g., a doctor’s note, police report, court notice, etc. clearly stating the date AND time of the mitigating problem. The student must also notify the CES Department Office/Instructor that the exam will be missed so that appropriate steps can be taken to make up the grade.

ADDITIONAL RESOURCES

Chemistry Tutoring Center: Located in the Central King Building, Lower Level, Rm. G12. Hours of operation are Monday - Friday 10:00 am - 6:00 pm. For further information please click [here](#).

Accommodation of Disabilities: Office of Accessibility Resources and Services (*formerly known as Disability Support Services*) offers long term and temporary accommodations for undergraduate, graduate and visiting students at NJIT.

If you are in need of accommodations due to a disability please contact Chantonette Lyles, Associate Director at the Office of Accessibility Resources and Services at 973-596-5417 or via email at lyles@njit.edu. The office is located in Fenster Hall Room 260. A Letter of Accommodation Eligibility from the Office of Accessibility Resources Services office authorizing your accommodations will be required.

For further information regarding self-identification, the submission of medical documentation and additional support services provided please visit the Accessibility Resources and Services (OARS) website at:

- <http://www5.njit.edu/studentuccess/disability-support-services/>

Important Dates See: Fall 2020 Academic Calendar, Registrar
<https://www5.njit.edu/registrar/fall-2020-academic-calendar/>

Date	Day	Event
September 1	T	First Day of Classes
September 5	S	Saturday Classes Begin
September 7	M	Labor Day

September 8	T	Monday Classes Meet Last Day to Add/Drop a Class Last Day for 100% Refund, Full or Partial Withdrawal
September 9	W	W Grades Posted for Course Withdrawals
September 14	M	Last Day for 90% Refund, Full or Partial Withdrawal No Refund for Partial Withdrawal after this date
September 28	M	Last Day for 50% Refund, Full Withdrawal
October 19	M	Last Day for 25% Refund, Full Withdrawal
November 9	M	Last Day to Withdraw
November 25	W	Friday Classes Meet
November 26	R	Thanksgiving Recess Begins
November 29	Su	Thanksgiving Recess Ends
December 10	R	Last Day of Classes
December 11	F	Reading Day 1
December 14	M	Reading Day 2
December 15	T	Final Exams Begin
December 21	M	Final Exams End
December 23	W	Final Grades Due

Course Outline

Lecture	Section	Topic	Assignment
1-2	1	Introduction and Review	
3-5	2	Structure and Properties of Organic Molecules	
6-7	3	Structure and Stereochemistry of Alkanes	
8-10	4	The Study of Chemical Reactions	
11-12	5	Stereochemistry	
13-15	6	Alkyl Halide, Nucleophilic Substitution and Elimination	
16-17	7	Structure and Synthesis of Alkenes	
18-19	8	Reactions of Alkenes	
20-21	9	Alkynes	
22-23	10	Structure and Synthesis of Alcohols	
24-25	11	Reactions of Alcohols	

