

Fall 2024

CHE 230-001: CHE Thermodynamics I

Xiaoyang XU

Follow this and additional works at: <https://digitalcommons.njit.edu/cme-syllabi>

Recommended Citation

XU, Xiaoyang, "CHE 230-001: CHE Thermodynamics I" (2024). *Chemical and Materials Engineering Syllabi*. 272.

<https://digitalcommons.njit.edu/cme-syllabi/272>

This Syllabus is brought to you for free and open access by the NJIT Syllabi at Digital Commons @ NJIT. It has been accepted for inclusion in Chemical and Materials Engineering Syllabi by an authorized administrator of Digital Commons @ NJIT. For more information, please contact digitalcommons@njit.edu.

Chem Engineer Thermodynamics I - CHE 230 - 001

Instructor: Professor Xiaoyang Xu
Departments of Chemical and Material Engineering
Office: Tiernan Hall, 362
Email: xiaoyang@njit.edu
Phone: 973-596-5359

Class hours:

Type	Time	Days	Where	Date Range	Schedule Type	Instructors
Class	1:00 pm- 3:05 pm	Tuesday Thursday	TIER- 411B	Sep 3, 2024 – Dec 23, 2024	Lecture	Xiaoyang Xu (P)

Office Hour: Tuesday 3:10 pm- 4:10 pm (Office: Tiernan Hall, 362); Please make appointments for in person meeting other than regular office hours.

TA and TA office hour: TBD

Office Hour: TBD (CME graduate office, Tiernan hall 324)

Estimated Workload: Lectures ~4 hours per week; quizzes/homework ~Please plan to spend a **minimum** of 10-12 hours per week on your homework problems for this course. Failure to meet this goal will seriously jeopardize your successful completion of this course and will harm your efforts in the junior and senior year.

Textbook: Smith, J. M., Van Ness, H. C., Abbott, M. M., & Swihart, M. T. “*Introduction to Chemical Engineering Thermodynamics*” 8th Edition, McGraw-Hill Education, New York, NY.

Supporting textbook: Donald P. Visco, Jr. Kevin D. Dahm "Fundamentals of Chemical Engineering Thermodynamics" ISBN: 9781111580711

Calculator: A calculator is required for solving problems.

Description: Thermodynamics is a science and, more importantly, an engineering tool used to describe processes that involve changes in temperature, transformation of energy, and the relationships between heat and work. The three introductory courses in the sophomore year, ChE 210, ChE230 and ChE240, and ChE 342 are the basic courses in chemical engineering fundamental principles. What you learn in these three courses will appear over and over again throughout your junior and senior courses. Therefore, it is in your best interest to learn these subjects well now.

Prerequisites: Chem 126, (or Chem 123), Math 112, Phys 111, (or Phys 106). Corequisite Math 211 (or Math 213). The Fundamentals of thermodynamics are applied to chemical engineering processes. Thermophysical properties and their engineering correlations are covered. Applications include chemical engineering and related fields such as environmental and biomedical engineering.

Course Goal: To help each student develop his/her problem-solving ability and gain insight into the process of Problem solving, with emphasis on thermodynamics. Specifically, this course is designed to help students learn to

- Apply conservation principles (mass and energy) to evaluate the performance of simple engineering systems and cycles,
- Evaluate thermodynamic properties of simple homogeneous substances,
- Analyze processes and cycles using the second law of thermodynamics to determine maximum efficiency and performance,
- Discuss the physical relevance of the numerical values for the solutions to specific engineering problems and the physical relevance of the problems in general,
- Evaluate the validity of the numerical solutions for specific engineering problems.

Topics to be covered:

1. Basic concepts; heat & work; steam tables
2. Energy balances in open and closed systems, including reacting systems
3. Thermodynamics quantities: enthalpy, entropy, internal energy, free energies
4. Steady state and unsteady-state processes
5. First and second laws of thermodynamics
6. Engines and power systems; Carnot and Rankine cycles
7. Turbines, pumps, and compressors, and refrigerations
8. Maxwell's relations; thermodynamic transformations
9. PVT behavior of ideal gases; equations of state

Lectures

- This course is a face-to-face course and attending the class sessions in person is mandatory.
- Food and drink are expressly prohibited in the classroom.
- Cellphones should be turned off during lectures.
- Students are expected to be in the classroom by the start time of each class.

Attendance: Attendance is mandatory. You must notify the instructor in advance if possible, of any absence by sending an email stating the date and reason for the absence. If you are absent for up to two class periods because of illness or injury, an email message stating the reason for absence will be sufficient. If you are absent from more classes because of illness or injury, verification of a visit to a health care professional may be required. Two times class absence (without verification/notification) will disqualify your final exam for this course.

Special Needs: If you need accommodations due to a documented disability please contact Office of the Dean of Students to discuss your specific needs. A Letter of Accommodation Eligibility authorizing your accommodations will be required.

Homework: Problems will be assigned. Homework will not be graded but similar problems will be tested in quiz, midterm exam or final exam. You are strongly recommended to work on homework by yourself and bring questions to lecture or office hour.

Course materials, office hours and correspondence

- The course Canvas page is the main platform for delivering information about the course.
- All relevant course materials and assignments will be posted on Canvas, so a student should check it regularly.
- Students must upload a professional-looking head shot for their Canvas profile.
- Students are strongly encouraged to attend office Hours. Long questions which require derivations will be discussed only during the office Hours and will not be answered by email.
- Questions regarding grades can be discussed only during the office Hours.
- E-mail and Canvas correspondence is intended only for quick questions. Questions which require a detailed discussion should be discussed in person during the office Hours.
- All correspondence should be conducted in a professional style, using formal English, having a greeting or a signature.
- To assure quick response to your emails, please add "ChE230" in the subject of your emails.

Examinations

There will be two 90 mins examinations during the term and a 150 mins final examination. The exams will be closed-notes and closed-book unless otherwise announced. Tutorial reviews will be held prior to each exam.

Quizzes: There will be quizzes occasionally at the beginning of the class. If you miss the class, you will miss the quiz that day. **There will be no makeup quiz!** Close book and close notes!

Policy on exams (other than final): A student must have a compelling reason to miss an exam. Documentation of the reason (e.g., doctor's note) is needed for the instructor to consider giving a make-up exam. A student who cannot make it to an exam needs to either e-mail or call and leave a voice message for the instructor **before** the exam is held. A student missing (for any reason) the first two exams has to withdraw from the course. A single (comprehensive) make-up exam will be given on the reading day (TBD) for those who have missed an exam for documented/ legitimate reasons.

Policy on final exam: The final exam will be based on the entire course material. Students missing the final exam without a documented serious excuse fail the course. Students missing the final exam with a documented serious reason get an Incomplete.

(Note: Quiz/exam answer should be clear and legible. Solution process should be given for calculation problems. **You may lose points if the writing is illegible or solution process is missing**)

Academic Dishonesty: Misrepresentation of a student's involvement in any required academic work will result in the instructor invoking the academic dishonesty policies of the university. This could result in an "F" grade being assigned for the course. Collaboration is expected for group activities, but not for individual assignments (such as exams). Instructions for each assignment should be followed. If in doubt, ask the instructor.

A calculator is allowed to solve the calculation problems. Any unauthorized electronic materials will be considered as a violation of academic integrity. The proctoring is subject to slight changes.

Grading:

Grades for the subject will be based on a total of 500 points:

ACTIVITIES	POINTS
-------------------	---------------

Two 90 mins midterms exams	100 (each)
----------------------------	------------

Quizzes	135
---------	-----

Final exam	165
------------	-----

Grading Scale (minimum cutoffs are firm):

85-100% A

80-84.9% B+

75-79.9% B

70-74.9% C+

65-69.9% C

55-64.9% D

<55% F

Quizzes and Exams grading

- Student handwriting must be legible in order to receive points.
- A student must show full details when solving a problem during an exam or a quiz. Not showing the work will cause the losing points even if the final answer is correct.
- Partial credit can be given for solving the exam and quiz problems, though no partial credit will be given if there are not enough details to follow.
- A student can dispute the exam scores within a week after the announcement of the score. Exam scores can only be disputed during the official Office Hours, not during class time or via email.

- A letter grade is based on the final score in accordance with the Tables given in this syllabus. The assigned letter grade is final and cannot be negotiated.
- Students will get zero for not coming to quizzes and there is no makeup quiz.
- Students will get zero for not coming to exams. If students miss an exam due to extreme circumstances (such as a medical problem), they need to notify the instructor via email before the beginning of the exam and bring proof of the circumstance to the Dean of Student's office. Only in the case of official approval from the Dean of Student's office, may a make-up be given at the discretion of the instructor.

2024 Fall Semester Class Schedule:

Sept 2	Labor Day. University Closed
Sept 3	First Day of Classes
Sept 9	Last Day to Add/Drop a Class
Sept 9	Last Day for 100% Refund, Full or Partial Withdrawal
Sept 10	W Grades Posted for Course Withdrawals
Sept 16	Last Day for 90% Refund, Full or Partial Withdrawal - No Refund for Partial Withdrawal after this date
Sept 30	Last Day for 50% Refund, Full Withdrawal
Oct 21	Last Day for 25% Refund, Full Withdrawal
Nov 11	Last Day to Withdraw from Classes
Nov 26	Thursday Classes Meet
Nov 27	Friday Classes Meet
Nov 28	Thanksgiving Recess Begins. No Classes
Dec 1	Thanksgiving Recess Ends
Dec 11	Last Day of Classes
Dec 12	Reading Day 1
Dec 13	Reading Day 2
Dec 14	Saturday Classes Meet
Dec 15	Final Exams Begin
Dec 21	Final Exams End
Dec 23	Final Grades Due