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## MTEN 610-101: Foundations of Materials Science and Engineering

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## **MTEN 610 – Foundations of Materials Science and Engineering**

Instructor: Dr. Joshua Young

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Office Hours: Tuesdays, 1 to 3 pm, or by appointment, York 322

Date, Time, and Location: Wednesdays, 6 to 9 pm, Faculty Memorial Hall 411

<u>**Textbook**</u>: Required: *Materials Science and Engineering: An Introduction, 9<sup>th</sup> Edition* – William D. Callister and David G. Rethwisch (2013), ISBN 9781118477700

This will also be supplemented with material from other textbooks and peer-reviewed literature. These books are not required, and any supplemental material will be provided to you. Material may come from: - *Materials Engineering: Bonding, Structure, and Structure-Property Relationships* – Susan Trolier-McKinstry, Robert E. Newnham

- Understanding Solids: The Science of Materials – Richard Tilley

- Physical Properties of Crystals: Their Representation by Tensors and Matrices – J. F. Nye

- Structure of Materials - Samuel M. Allen, Edwin L. Thomas

<u>Course Overview</u>: This is an introductory course for Materials Engineering graduate students. This course will introduce materials from an atomistic point of view and describe the relationship between their underlying structure and properties. We will first cover chemical bonding and crystal structure, and how this results in differences between families of materials (metals, ceramics, and polymers). We will then discuss phase transformations and use phase diagrams to describe the compositions and microstructure of solid solutions. Finally, we will discuss important properties of materials, including electronic, magnetic, optical, and thermal, and how they are selected and used in various modern engineering applications.

**Prerequisites**: introductory thermodynamics; modern physics; general chemistry; calculus

**<u>Course Objectives</u>**: At the end of this course, students should be able to understand:

- (1) the fundamental link between materials' structure, processing, and properties
- (2) the differences between families of materials
- (3) how to use phase diagrams to describe transformations and microstructure of solid solutions
- (4) how the properties of materials influence their use in engineering applications

<u>Attendance Policy</u>: Attendance is expected at all classes. In lieu of formal attendance, however, participation in in-class activities will count towards your final course overall grade (see below).

<u>Academic Integrity</u>: Academic Integrity is the cornerstone of higher education and is central to the ideals of this course and the university. Cheating is strictly prohibited and devalues the degree that you are working on. As a member of the NJIT community, it is your responsibility to protect your educational investment by knowing and following the academic code of integrity policy that is found at: <u>http://www5.njit.edu/policies/sites/policies/files/academic-integrity-code.pdf</u>.

Please note that it is my professional obligation and responsibility to report any academic misconduct to the Dean of Students Office. Any student found in violation of the code by cheating, plagiarizing or using any online software inappropriately will result in disciplinary action. This may include a failing grade of F,

and/or suspension or dismissal from the university. If you have any questions about the code of Academic Integrity, please contact the Dean of Students Office at dos@njit.edu.

**<u>Grading</u>**: The final grade for the course is divided as follows:

- Homework Sets = 10% of grade
- In-Class Exercise/Participation = 10% of grade
- Exam 1 = 30% of grade
- Exam 2 = 30% of grade
- Final Project = 20% of grade

Grades will be assigned with the following rubric:

90% and above	А
85-89%	B+
80-84%	В
75-79%	C+
70-74%	С
60-69%	D
Below 60%	F

<u>Homework:</u> Students will be assigned 8 homework sets of exercises, each equally weighted. Homework sets can be turned in electronically formatted and printed *or* legibly handwritten at the start of the class in which they are due or emailed to the instructor *before* class begins. Homework turned in after 6 pm on the due date will be automatically reduced by 50%. *You should begin problem sets early*. Working with your peers on the homework is allowed and encouraged. You should acknowledge your collaborators when submitting your assignments. Each student must provide their own solutions to problems and submit their own assignment, with their own written answers to the questions. It is not acceptable (and a violation of course policies) to use the same written answers as your collaborators.

<u>In-Class Exercises</u>: There will be in class exercises, both individual and in small groups, covering material presented during the lecture. This can be group problem solving, short quizzes, brief writing assignments, or other activities.

<u>Exams</u>: There will be two exams during the semester. The exams will be based on in-class material and homework problems. The midterm exam will cover all material from weeks 1 through 4; the final exam will be cumulative, but with a specific focus on topics covered after the midterm. There will be no makeup midterm or final exams baring extreme extenuating circumstances.

<u>Final Project</u>: The final project will involve selecting an engineering application, researching the materials used for it, and giving a 15 minute presentation to the class about your findings. The presentation will be in the style of a Materials Research Society meeting, with 12 minutes for the presentation and 3 minutes for questions. Potential topics and further details will be provided the week after the midterm exam.

Week	Topics	Assignment
1 – Sep. 4, 2019	- Course Objectives	
	- Introduction	
	- Atomic Structure and Bonding	
2 – Sep. 11, 2019	- Atomic Structure and Bonding	

## Schedule of Topics:

	- Crystal Structure Basics	
3 – Sep. 18, 2019	- Advanced Crystal Structure Topics	HW1 due
	- Symmetry and Tensor Properties	
4 – Sep. 25, 2019	- Electronic Structure	HW2 due
	- Metals and Ceramics	
5 – Oct. 2, 2019	- Defects	HW3 due
	- Diffusion	
6 – Oct. 9, 2019	- Polymers	HW4 due
	- Molecular Solids	
	- Review for Midterm	
7 – Oct. 16, 2019	Midterm Exam	
8 – Oct. 23, 2019	- Phase Transformations I: Phase Diagrams and	
	Microstructure	
	- Project Topics assigned	
9 – Oct. 30, 2019	- Phase Transformations II: Solidification, Nucleation	HW5 due
	and Growth	
10 – Nov. 6, 2019	- Mechanical Properties	HW6 due
	- Dislocations	
	- Failure	
11 – Nov. 13, 2019	- Electronic Properties	HW7 due
	- Devices	
12 – Nov. 20, 2019	- Magnetic Properties	HW8 due
	- Optical Properties	
13 – Nov. 27, 2019	NO CLASS – FRIDAY CLASSES MEET	
14 – Dec. 4, 2019	- Student Presentations and Review for Final Exam	
15 – Dec. 11, 2019	- Student Presentations and Review for Final Exam	
16 – Dec. 18, 2019	Final Exam	