

Fall 2018

CHEM 360 - Environmental Chemistry

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CHEM 360, Environmental Chemistry

Fall 2018 Course Syllabus

NJIT Academic Integrity Code: All Students should be aware that the Department of Chemistry & Environmental Science (CES) takes the University Code on Academic Integrity at NJIT very seriously and enforces it strictly. This means that there must not be any forms of plagiarism, i.e., copying of homework, class projects, or lab assignments, or any form of cheating in quizzes and exams. Under the University Code on Academic Integrity, students are obligated to report any such activities to the Instructor.

COURSE INFORMATION

Course Description: Chemistry of the environment with emphasis on the atmosphere. Included are an introduction to the composition and chemistry of the natural and polluted atmosphere, thermodynamics and kinetics of atmospheric reactions, indoor and outdoor air pollution, air quality and its impact on human health, air quality regulations, and climate change. Examples of specific environmental issues covered in this course are the global stratospheric ozone depletion and Polar ozone holes, classical and photochemical smog, acid rain, and climate warming.

Number of Credits: 3

Prerequisites: CHEM 126 or CHEM 122, or CHEM 124 with a grade of C or better. CHEM 360 is a prerequisite for CHEM 361.

Course-Section and Instructors

Course-Section	Instructor
001	Alexei Khalizov

Office Hours for All Chemistry & Environmental Science Instructors: [Fall 2018 Office Hours and Emails](#)

Required Textbook:

Title	Atmospheric Chemistry (From the Surface to the Stratosphere)
Author	Grant Ritchie
Edition	1st
Publisher	Oxford
ISBN #	978-1-78634-176-1

Supplementary textbook (not required, but highly recommended): Elements of Environmental Chemistry by Ronald A. Hites, 2nd Edition (available electronically via NJIT library website at <http://librarius.njit.edu/vwebv/holdingsInfo?bibId=423296>).

General Chemistry reference materials (not required): (a) ACS General Chemistry Study Guide (<http://uwm.edu/acs-exams/instructors/ordering-information/>); (b) a good freshman General Chemistry

textbook, such as Chemistry: a Molecular Approach by N.J. Tro (any edition)

Calculator requirements: bring to every class scientific or engineering calculator. Advanced graphing calculators are NOT allowed during exams and quizzes (e.g., TI-30 or TI-34 are OK, but not TI-84 or TI-Nspire)

University-wide Withdrawal Date: The last day to withdraw with a W is Monday, November 12, 2018. It will be strictly enforced.

Learning Outcomes: by the end of this course, students will be able to

- calculate concentrations and mixing ratios of pollutants using different units
- describe the concepts of global cycles, sources and sinks, and lifetimes of pollutants
- calculate lifetimes and removal rates of pollutants
- identify primary and secondary pollutants
- name criteria pollutants and justify their selection by the EPA
- calculate pH of rainwater under natural and polluted conditions
- describe pollution control methods, regulations, and policies
- assess impacts of air pollution on the environment, human health, and climate
- explain the major differences between the successive layers of the earth's atmosphere
- describe and explain the major photochemical reactions taking place in the stratosphere
- list the photochemical reactions leading to the ozone depletion
- list the chemical reactions forming photochemical smog
- describe the greenhouse effect and its connection to global warming
- explain the phenomena behind the formation of acid rain

POLICIES

All CES students must familiarize themselves with, and adhere to, all official university-wide student policies. CES takes these policies very seriously and enforces them strictly.

Grading Policy: The final grade in this course will be determined as follows:

Homework	See below
Quizzes	30%
Class project	10%
Midterm Exam	25%
Final Exam	30%

Your final letter grade in this course will be based on the following tentative curve:

A	90+	C	70+
B+	85+	D	60+
B	80+	F	
C+	75+		

Attendance Policy: Attendance at classes will be recorded and is **mandatory**. Each class is a learning experience that cannot be replicated through simply "getting the notes."

Homework/Quiz Policy: Each homework assignment is followed by a quiz. The grade for each quiz is counted towards the total grade only if the associated homework assignment is turned in on time. The quiz with the lowest grade will not be counted towards total.

Exams: There will be one midterm exam held in class during the semester and one comprehensive final exam. The following exam periods are tentative and therefore possibly subject to change:

Midterm Exam	Mid October
Final Exam Period	December 15 - 21, 2018

The final exam will test your knowledge of all the course material taught in the entire course.

Makeup Exam Policy: There will normally be **NO MAKE-UP QUIZZES OR EXAMS** during the semester. In the event that a student has a legitimate reason for missing a quiz or exam, the student should contact the Dean of Students office and present written verifiable proof of the reason for missing the exam, e.g., a doctor's note, police report, court notice, etc. clearly stating the date AND time of the mitigating problem. The student must also notify the CES Department Office/Instructor that the exam will be missed so that appropriate steps can be taken to make up the grade.

Cellular Phones: All cellular phones and other electronic devices must be switched off during all class times. Such devices must be stowed in bags during exams or quizzes.

Class project: may be based on (a) scientific article from a peer-reviewed journal, (b) your research project (e.g., independent study), or (c) development of a demonstration for CHEM360/361. The article **must cover one of the topics from this course** and **cannot be a magazine article or a review article**. Students will work individually and must present the original journal article of their choice, a one-paragraph summary of their research project, or a one-paragraph demonstration proposal (including the source, if available) for approval by the instructor by **September 25**. Presentations (10 min) will begin on **October 9** and will continue throughout the semester (~2-3 presentations each week). The summary will be due on **October 30** (1 page, single-spaced, submitted electronically via Moodle, will be checked for plagiarism using Turnitin).

ADDITIONAL RESOURCES

Chemistry Tutoring Center: Located in the Central King Building, Lower Level, Rm. G12. Hours of operation are Monday - Friday 10:00 am - 6:00 pm. For further information please click [here](#).

Accommodation of Disabilities: Office of Accessibility Resources and Services (**formerly known as Disability Support Services**) offers long term and temporary accommodations for undergraduate, graduate and visiting students at NJIT.

If you are in need of accommodations due to a disability please contact Chantonette Lyles, Associate Director at the Office of Accessibility Resources and Services at **973-596-5417** or via email at lyles@njit.edu. The office is located in Fenster Hall Room 260. A Letter of Accommodation Eligibility from the Office of Accessibility Resources Services office authorizing your accommodations will be required.

For further information regarding self-identification, the submission of medical documentation and additional support services provided please visit the Accessibility Resources and Services (OARS) website at:

- <http://www5.njit.edu/studentssuccess/disability-support-services/>

Important Dates (See: [Fall 2018 Academic Calendar, Registrar](#))

Date	Day	Event
September 4, 2018	T	First Day of Classes
September 10, 2018	M	Last Day to Add/Drop Classes
November 12, 2018	M	Last Day to Withdraw
November 20, 2018	T	Thursday Classes Meet
November 21, 2018	W	Friday Classes Meet
November 22 - 25, 2018	R - Su	Thanksgiving Break - University Closed

December 12, 2018	W	Last Day of Classes
December 14, 2018	F	Reading Day
December 15 - 21, 2018	F - R	Final Exam Period

Course Outline

Lecture	Topic	Assignment
1	Introduction: environment; course logistics; Earth as a closed system (lithosphere, atmosphere, and hydrosphere); life and ecosystem; global biogeochemical cycles	RAH 1.2
2	Review of fundamental concepts #1: concentration units for gas mixtures and aqueous solutions; unit conversions; peer reviewed journal articles and literature search	RAH 1
3	Earth's atmosphere: major and minor constituents; atmospheric structure; energy balance; global circulation; sources, sinks, transport, and lifetimes of chemicals	GR 2.1, 2.3.1, Appendix A; RAH 3.1
4	Review of fundamental concepts #2: atoms and molecules; structure and reactivity; thermochemistry; equilibria; reaction rates; pH of strong and weak acids; photochemistry; heterogeneous reactions	GR 1.6, 5.5; RAH 2.3, 3.2, 5.1, Appendix A
5	Atmospheric aerosols: physical properties and chemical composition; sources and sinks; interaction with sunlight; aerosol-cloud interaction	GR 5.1-5.4
6	Climate change: radiative balance; greenhouse gases; signs of climate change; radiative forcing; global warming and its impacts	GR 2.3, 2.4; RAH 4
7	Stratospheric chemistry - ozone chemistry: Chapman model of stratospheric ozone; catalytic ozone destruction; global stratospheric ozone reduction; Montreal protocol; Antarctic ozone hole	GR 3; RAH 3.4, 3.5
8	Tropospheric chemistry: the hydroxyl radical; oxidation of methane and carbon monoxide; the role of nitrogen oxides and organic compounds in the formation of tropospheric ozone and secondary organic aerosols; oxidation of sulfur dioxide; nighttime chemistry	GR 4; RAH 3.6
9	Air quality: indoor and outdoor air pollution; classical and photochemical smog; primary and secondary pollutants; criteria pollutants; monitoring of air pollutants; health impacts of air pollution; air pollution regulations; acid rain	GR 4.7, 5.6; RAH 3.6, 5.1, 5.2

*Updated by Alexei Khalizov - 2018
Department of Chemistry & Environmental Sciences
Course Syllabus, Fall 2018*