

Fall 2018

PHEN 501 - Pharmaceutical Engineering Fundamentals II

Boris Khusid

Follow this and additional works at: <https://digitalcommons.njit.edu/cme-syllabi>

Recommended Citation

Khusid, Boris, "PHEN 501 - Pharmaceutical Engineering Fundamentals II" (2018). *Chemical and Materials Engineering Syllabi*. 13.
<https://digitalcommons.njit.edu/cme-syllabi/13>

This Syllabus is brought to you for free and open access by the NJIT Syllabi at Digital Commons @ NJIT. It has been accepted for inclusion in Chemical and Materials Engineering Syllabi by an authorized administrator of Digital Commons @ NJIT. For more information, please contact digitalcommons@njit.edu.

New Jersey Institute of Technology
Otto H. York Department of Chemical & Materials Engineering
Fall 2018

Course: PhEn 501 Pharmaceutical Engineering Fundamentals II (3 credits)

Prerequisites: Undergraduate courses in chemistry and mathematics and, if needed, PhEn 500 that can also be taken concurrently with this course.

Reason for course prerequisites: Undergraduate courses in chemistry covers states and properties of matter, chemical equilibrium, thermodynamics and its application to chemical and biochemical systems. Undergraduate courses in mathematics and PhEn 500, if needed, provide a mathematical background, including differential equations, required for quantification of chemical and pharmaceutical processes.

Days/ Times: Thursday: 6:00PM – 9:05PM

Room: GITC 1403

Instructor: Dr. Boris Khusid

Professor, Otto H. York Department of Chemical Engineering

FMH 215, Tel: 973-596-3316, Fax: 973-596-8436, E-mail: khusid@njit.edu

<http://chemicaleng.njit.edu/people/khusid.php>

Office Hours Tuesday: 9:30AM – 11:30AM

Note: you can always schedule an appointment by email if the office hour time conflicts with your classes

Course description: This bridge course is required for students who are admitted to MS programs in Pharmaceutical Engineering or Chemical Engineering without an undergraduate engineering degree or with an engineering background that did not include the topics covered by this course. The course covers the fundamentals of chemical engineering calculations based on material and energy balances on chemical processes; estimation of physical properties of substances, phase and reaction equilibrium in multi-component systems; and chemical kinetics in reactive processes. It provides the foundation for subsequent courses in pharmaceutical and chemical engineering and is not counted toward degree credit related to MS programs in Pharmaceutical Engineering and Chemical Engineering.

The course combines lectures & problem-solving to provide students with the ability to do the following:

- Formulate and solve material and energy balance equations for basic processes in chemical and pharmaceutical industry
- Communicate effectively the acquired knowledge in written & verbal form

Required textbook: Elementary Principles of Chemical Processes, 4th Ed., Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard July 2015 © 2016, John Wiley & Sons

<http://www.wiley.com/WileyCDA/WileyTitle/productCd-EHEP003498.html>

Course objectives: Use material and energy balance theory to design and analyze basic chemical and pharmaceutical processes:

- Material Balances: Process flowchart, law of conservation of mass, degree-of-freedom analysis, estimation of physical properties of substances, material balance calculations
- Energy Balances: Law of conservation of energy, closed and open process systems, estimation of thermodynamic properties of substances, energy balance calculations
- Reactive Processes: Material and energy balances on reactive processes, exothermic and endothermic reactions, extent of reaction

Measurable outcomes:

- Proficiency in using the basic principles of material and energy balance theory to analyze processes designed to transform raw materials into desired products

Course outline and tentative weekly listing (15-week schedule)

Topic	Week	Outline	Book Chapter
<i>Engineering Calculations and Processes Variables</i>	1	Units and dimensions; sample mean, range, sample variance and sample standard deviation, method of least squares; process variables: temperature, pressure, chemical composition	2, 3
<i>Fundamentals of Material Balances</i>	2	Process classification, flowchart, general differential and integral balances, material balance calculations	4
<i>Fundamentals of Material Balances</i>	3	Material balances on single-unit processes, degree-of-freedom analysis	4
<i>Fundamentals of Material Balances</i>	4	Material balances on multiple-unit processes, recycle and bypass processes, degree-of-freedom analysis	4
<i>Fundamentals of Material Balances</i>	5	Chemical reaction stoichiometry, limiting and excess reactants, multiple reactions, material balances on reactive processes	4
<i>Single-Phase Systems</i>	6	Liquids, ideal gases, equations of state for non-ideal gases, gas mixtures	5
<i>Multiphase Systems</i>	7	Phase-change processes, the Gibbs phase rules, gas-liquid systems	6
<i>Multiphase Systems</i>	8	Multi-component gas-liquid systems, liquid-liquid systems, liquid-solid systems	6
Midterm Exam 1	9		
<i>Energy Balances</i>	10	Forms of energy, energy balances on open systems, thermodynamic data	7
<i>Balances on Non-reactive Processes</i>	11	Energy balance calculations for open and closed systems, processes involving phase changes	8
Midterm Exam 2	12		
<i>Balances on Reactive Processes</i>	13	Heats of reaction, heats of combustion, energy balances on reactive systems	9
<i>Balances on Transient Processes</i>	14	Differential material and energy balances, single-phase nonreactive processes	11
Final Exam	15		

Homework assignment, format guidelines, and grading

- Homework is collected at the beginning of each lecture
- Late homework will not be accepted for grading
- Feedback on the homework will be provided during lectures, solutions will be discussed, and graded homework will be returned
- Each problem will be graded individually
- Structure the solution into the following sections:
 - Known* - The problem is posed
 - Find* - The quantities to be found are stated

Sketch - The flowchart and process variables are presented
Properties The physical properties/data needed to solve the problem are listed
Analysis - The problem is solved in a systematic manner, showing all steps, the fundamental equations from which the calculation begins are included, and all numerical values (including units) are shown

Discussion - Comments are made on the results, as appropriate

- Arrange problems in numerical order
- Print your name at the top of each page
- Write only on 8½" x 11" paper; start each problem on a new page
- Staple all pages together and bring a hard copy to class or email scanned sheets or files in .doc or .pdf formats to Instructor with Cc to Teaching Assistant.

Midterm and final exams

- There will be open-book and open-lecture notes midterm exams. Exact date of midterm exams will be announced a week before.
- An open-book and open- lecture notes comprehensive final exam will cover all material in the course.
- The midterm and final exams must be completed individually, in accordance with the NJIT Honor Code.
- Each problem on the midterm and final exams will be graded individually.

A missed midterm exam will be averaged into the final grade as *zero*, unless permission is obtained from the instructor prior to the exam. Permissions are granted only for very serious circumstances attested to by the NJIT administration, verifiable and significant medical problems, religious holidays, and also serious personal situations, such as deaths in the family. A student who has been excused will be required to take a makeup exam.

- Exams will be administered either at NJIT or at testing centers, as described below:
 - According to NJIT policy, all students taking the class but living within 50 miles of NJIT must take the required exams at NJIT's main campus in Newark, NJ. The exact time and the room where the exam will take place will be specified by the instructor prior to the exam;
 - Students living outside this 50-mile zone can only take the exams at predefined testing locations. The National College Testing Association (NCTA) lists the participating institutions nationwide (<http://www.ncta-testing.org/list-of-certified-centers>) where students can take proctored tests according to the rules set up by the course instructor. Students should visit this website, identify the testing location where they plan to take the exams, and inform the course instructor within two weeks from the beginning of classes of which testing location they have selected. Students should be aware that they will be required to pay a fee to the chosen testing center for each exam that they take there. Such a fee is typically on the order of \$25-\$50 per exam, depending on the testing center. Prior to each exam, students should make arrangements with the selected testing center to ensure that they can take the exam at the predefined location at a given day and time.

Assessment criteria and grading

The course has been designed so that lectures, homework assignments, midterm and final exams are integral and essential parts of the learning process. Final grades will be determined from scores as follows:

Midterm exam 1: 25% Midterm exam 2: 25% Homework: 20% Final Exam: 30%

The final grade will be assigned on the basis of "*a curve*".

Course materials

Textbook, lecture notes, relevant publications and websites

Accommodations due to disability

If you need accommodations due to a disability please contact Chantonette Lyles, Associate Director of Disability Support Services, Fenster Hall Room 260 to discuss your specific needs. A Letter of Accommodation Eligibility from the Disability Support Services office authorizing your accommodations will be required